

## 1 Nonproprietary Names

PhEur: Rapeseed Oil, Refined  
USP-NF: Canola Oil

## 2 Synonyms

Canbra oil; *Colza* CT; huile de colza; *Lipex 108*; *Lipex 204*; *Lipovol* CAN; low erucic acid colza oil; low erucic acid rapeseed oil; rapae oleum raffinatum; tower rapeseed oil.

## 3 Chemical Name and CAS Registry Number

Canola oil [120962-03-0]

## 4 Empirical Formula and Molecular Weight

Canola oil contains approximately 6% saturated acids, 2% monounsaturated acids, and 32% polyunsaturated acids. Additionally, sulfur-containing fatty acids may also be present as minor constituents.

The sulfur-containing compounds have been held responsible for the unpleasant odors from heated rapeseed oil. It has been suggested that the sulfur compounds in rapeseed oil are of three types: volatile, thermolabile, and nonvolatile.<sup>(1)</sup>

Unrefined canola oil is said to contain low levels of sulfur-containing fatty acids, resulting in the presence of sulfur in the oil in the stable form of triglycerides. These triglycerides resist refining procedures.<sup>(2)</sup> See Table I for the sulfur content of crude, refined, and deodorized canola oils.<sup>(3)</sup>

**Table I:** Total sulfur content in crude, refined and bleached and deodorized canola oil.<sup>(a)</sup>

Oil sample	Range (mg/kg)	Mean	Standard deviation
Crude	23.6–24.1	23.8	1.0
Refined	19.1–20.2	19.7	2.85
Bleached and deodorized	15.6–16.5	16.2	2.7

(a) Determined using five replicates of each sample analyzed by ion chromatography.

## 5 Structural Formula

See Section 4.

## 6 Functional Category

Emollient; lubricant; oleaginous vehicle.

## 7 Applications in Pharmaceutical Formulation or Technology

Canola oil is a refined rapeseed oil obtained from particular species of rapeseed that have been genetically selected for their low erucic acid content.<sup>(4)</sup> In pharmaceutical formulations, canola oil is used mainly in topical preparations such as soft soaps and liniments. It is also used in cosmetics.

## 8 Description

A clear, light yellow-colored oily liquid with a bland taste.

## 9 Pharmacopeial Specifications

See Table II.

**Table II:** Pharmacopeial specifications for canola oil.

Test	PhEur 6.2	USP32–NF27
Identification	+	+
Characters	+	–
Specific gravity	–	0.906–0.920
Acid value	≤ 0.5	≤ 6.0
Alkaline impurities	+	–
Iodine value	–	110–126
Peroxide value	≤ 10.0	≤ 10.0
Saponification value	–	178–193
Unsaponifiable matter	≤ 1.5%	≤ 1.5%
Refractive index	–	1.465–1.467
Heavy metals	–	0.001%
Fatty acid composition	+	+
Carbon chain length < 14	–	0.1%
Eicosenoic acid	≤ 5.0%	< 2.0%
Erucic acid	≤ 2.0%	≤ 2.0%
Linoleic acid	16.0–30.0%	< 40%
Linolenic acid	6.0–14.0%	< 14%
Oleic acid	50.0–67.0%	> 50%
Palmitic acid	2.5–6.0%	< 6.0%
Stearic acid	≤ 3.0%	< 2.5%

## 10 Typical Properties

**Density** 0.913–0.917 g/cm<sup>3</sup>

**Flash point** 290–330°C

**Free fatty acid** ≤ 0.05% as oleic acid

**Freezing point** –10 to –2°C

**Solubility** Soluble in chloroform and ether; practically insoluble in ethanol (95%); miscible with fixed oils.

**Viscosity (dynamic)** 77.3–78.3 mPa s (77.3–78.3 cP) at 20°C

## 11 Stability and Storage Conditions

Canola oil is stable and should be stored in an airtight, light-resistant container in a cool, dry place. The USP32–NF27 specifies that contact between canola oil and metals should be avoided. Containers should be filled to the top, while partially filled containers should be flushed with nitrogen. During storage, grassy, paintlike, or rancid off-flavors can develop.

Flavor deterioration has been attributed mainly to secondary oxidation products of linolenic acid, which normally makes up 6–14% of the fatty acids in canola oil. Storage tests of canola oil showed sensory changes after 2–4 days at 60–65°C in comparison to 16 weeks at room temperature. Canola oil seems to be more stable to storage in light than cottonseed oil and soybean oils, but is less stable than sunflower oil.<sup>(5)</sup> In addition, the effects of various factors on sediment formation in canola oil have been reported.<sup>(6)</sup>

It has been reported that oils stored at 2°C showed the highest rate of sediment formation, followed by those stored at 6°C.<sup>(5)</sup> All samples showed little sediment formation, as measured by turbidity, during storage at 12°C. Removal of sediment from canola oil prior to storage by cold precipitation and filtration did not eliminate this phenomenon, which still developed rapidly at 2°C.

A study on the effect of heating on the oxidation of low linolenic acid canola oil at frying temperatures under nitrogen and air clearly showed that a significantly lower development of oxidation was evident for the low linolenic acid canola oil. Reduction in the linolenic acid content of canola oil reduced the development of room odor at frying temperatures.

The thermal oxidation of canola oil studied during oven heating revealed an increase in peroxide values of pure and antioxidant-

treated oils. Peroxide values were shown to differ between pure and antioxidant-treated canola oil during the initial stages of microwave heating (6 minutes). Formation of secondary products of oxidation, which contribute to off-flavors, were also observed.<sup>(7)</sup>

## 12 Incompatibilities

## 13 Method of Manufacture

Canola oil is obtained by mechanical expression or *n*-hexane extraction from the seeds of *Brassica napus* (*Brassica campestris*) var. *oleifera* and certain other species of *Brassica* (Cruciferae). The crude oil thus obtained is refined, bleached, and deodorized to substantially remove free fatty acids, phospholipids, color, odor and flavor components, and miscellaneous nonoil materials.

## 14 Safety

Canola oil is generally regarded as an essentially nontoxic and nonirritant material, and has been accepted by the FDA for use in cosmetics, foods, and pharmaceuticals.

Rapeseed oil has been used for a number of years in food applications as a cheap alternative to olive oil. However, there are large amounts of erucic acid and glucosinolates in conventional rapeseed oil, both substances being toxic to humans and animals.<sup>(7)</sup> Canola oil derived from genetically selected rapeseed plants that are low in erucic acid content has been developed to overcome this problem. The FDA specifies 165.55 mg as the maximum amount for each route or dosage form containing the ingredient.

Feeding studies in rats have suggested that canola oil is nontoxic to the heart, although it has also been suggested that the toxicological data may be unclear.<sup>(8)</sup>

## 15 Handling Precautions

Observe normal precautions appropriate to the circumstances and quantity of material handled. Spillages of this material are very slippery and should be covered with an inert absorbent material prior to disposal. Canola oil poses a slight fire hazard.

## 16 Regulatory Status

Accepted for use by the FDA in cosmetics and foods. Included in the FDA Inactive Ingredients Database (oral capsules). Included in the Canadian List of Acceptable Non-medicinal Ingredients.

## 17 Related Substances

Almond oil; corn oil; cottonseed oil; peanut oil; rapeseed oil; sesame oil; soybean oil.

### Rapeseed oil

CAS number [8002-13-9]

Synonyms Calchem H-102; colza oil; rape oil.

Appearance A clear, yellow to dark yellow-colored oily liquid.

Iodine number 94–120

Peroxide value <5

Saponification value 168–181

Comments Rapeseed oil contains 40–55% erucic acid. It is an edible oil and has been primarily used as an alternative, in foods and some pharmaceutical applications, to the more expensive olive oil. However, the safety of rapeseed oil as part of the diet has been questioned; see Section 14.

## 18 Comments

Canola oil has the lowest level of saturated fat compared to all other oils on the market at present and it is now second only to soybean as the most important source of vegetable oil in the world. It has both a

high protein (28%) and a high oil content (40%). When the oil is extracted, a high-quality and highly palatable feed concentrate of 37% protein remains. Canola oil is also high in the monounsaturated fatty acid oleic acid; see Table III.

The content of tocopherol, a natural antioxidant in canola, is comparable to those of peanut and palm oil. This is an important factor for oils with high linolenic acid content, which can reduce the shelf-life of the product, while the natural antioxidant, if present, can prevent oxidation during storage and processing.

A specification for canola oil is contained in the Food Chemicals Codex (FCC).<sup>(10)</sup>

The EINECS number for canola oil is 232-313-5.

**Table III:** Comparison of the composition of crude soybean, canola, palm, and peanut oils.

Components	Canola	Palm	Peanut	Soybean
Fatty acid (%)	0.4–1.0	4.6	0.5–1.0	0.3–0.7
Phosphatides (gum) (%)	3.6	0.05–0.1	0.3–0.4	1.2–1.5
Sterols/triterpene alcohol (%)	0.53	0.1–0.5	0.2	0.33
Tocopherols (%)	0.06	0.003–0.1	0.02–0.06	0.15–0.21
Carotenoids (mg/kg)	25–50	500–1600	>1	40–50
Chlorophyll/pheophytins (ppm)	5–25	—	—	1–2
Sulfur (ppm)	—	—	—	12–17
Iodine value	112–131	44–60	84–100	123–139

## 19 Specific References

- 1 Devinat G *et al.* Sulfur-compounds in the rapeseed oils. *Rev Fr Corps Gras* 1980; 27: 229–236.
- 2 Wijesundera RC, Ackman RG. Evidence for the probable presence of sulfur-containing fatty-acids as minor constituents in canola oil. *J Am Oil Chem Soc* 1988; 65: 959–963.
- 3 Abraham V, de Man JM. Determination of total sulfur in canola oil. *J Am Oil Chem Soc* 1987; 64: 384–387.
- 4 Hiltunen R *et al.* Breeding of a zero erucic spring turnip-rape cultivar, *Brassica campestris* L. adapted to Finnish climatic conditions. *Acta Pharm Fenn* 1979; 88: 31–34.
- 5 Przybylski R *et al.* Formation and partial characterization of canola. *J Am Oil Chem Soc* 1993; 70: 1009–1016.
- 6 Liu H *et al.* Effects of crystallization conditions on sediment. *J Am Oil Chem Soc* 1994; 71: 409–418.
- 7 Vieira T *et al.* Canola oil thermal oxidation during oven test and microwave heating. *Lebensm-Wiss Technol* 2001; 34: 215–221.
- 8 Anonymous. Rapeseed oil revisited. *Lancet* 1974; ii: 1359–1360.
- 9 Anonymous. Rapeseed oil and the heart. *Lancet* 1973; ii: 193.
- 10 *Food Chemicals Codex*, 6th edn. Bethesda, MD: United States Pharmacopeia, 2008; 156.

## 20 General References

- Koseoglu SS, Iusas EW. Recent advances in canola oil hydrogenations. *J Am Oil Chem Soc* 1990; 67: 3947.
- Malcolmson LJ *et al.* Sensory stability of canola oil: present status. *J Am Oil Chem Soc* 1994; 71: 435–440.
- Raymer PL. Canola: an emerging oilseed crop. Janick J, Whipkey A, eds. *Trends in New Crops and New Uses*. Alexandria, VA: ASHS Press, 2002; 122–126.

## 21 Authors

KS Alexander, R Milallos.

## 22 Date of Revision

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